## STEROID SAPONINS AND SAPOGENINS

OF Allium

VI. THE 3-O- $\beta$ -D-GLUCOPYRANOSIDE OF KARATAVIGENIN B

were fragments of benzoic acid with m/e 122 ( $C_7H_6O_2$ ) and 105 ( $C_7H_5O$ ).

F. S. Khristulas, M. B. Gorovits, and N. K. Abubakirov

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We have continued a study of all the extracted substances of steroid nature isolated previously [1] from the bulbs of Allium karataviense Rgl. (family Alliaceae). By chromatography of the extract on  $SiO_2$  [with elution by chloroform—methanol (25:1)] we have isolated 0.33% (on the air-dry weight of the raw material) of a glycoside (IV),  $C_{40}H_{58}O_{12}$ , mp 294-296°C (ethyl acetate);  $[\alpha]_D^{20}-98.1\pm2^\circ$  (c 1.64; pyridine);  $\lambda \frac{C_2H_5OH}{max}$  230 nm (log  $\epsilon$  4.20);  $\nu$  kBr 3400-3580 (OH); 1710, 1280 (ester group); 870, 900>930, 995 (spiroketal chain, R series) [2, 3]; 1605, 1585, and 720 cm<sup>-1</sup> (benzene ring).

The alkaline hydrolysis of compound (IV) in a 1% methanolic solution of KOH gave a glucoside (II) identified as alliogenin glucopyranoside [1, 4]. Benzoic acid was found in the hydrolyzate by TLC. The acid hydrolysis of (IV) with 5% of HCl in methanol at the boil for 2.5 h led to the production of a mixture of genins (I) and (III) separated by chromatography on silica gel. Elution by chloroform—methanol (100:1) gave the genin (III), which we have called karatavigenin B,  $C_{34}H_{48}O_7$ , mp 301-303°C (methanol),  $[\alpha]_{20}^{20} = 84.7 \pm 2^\circ$  (c 1.18 pyridine);  $\lambda C_2H_5OH$  230 nm (log  $\epsilon$  4.14);  $\lambda KBr$  3400-3500 (OH); 1695, 1280 (ester grouping); 870, 903 > 930; 990 (spiroketal chain R series); 1603, 1590, 720 cm<sup>-1</sup> (benzene ring). In the mass spectrum of karatavigenin B (170°C, 40 eV), in addition to the molecular ion with m/e 568 and the peaks of ions with m/e 509, 499, 496, 454, 439, 42 ,\*139, and 115, which are characteristic for the fragmentation of steroid sapogenins [5], there

The genin (1), eluted with chloroform—methanol (20:1), was identified as alliogenin [4].

$$\begin{array}{l} I \ R := R_1 = H \\ II \ R := H; \ R_1 = C_0 H_{11} O_5 \ (\text{D-glucose residue}) \\ III \ R = C_0 H_5 CO; \ R_1 = H. \\ IV \ R = C_0 H_5 CO; \ R_1 = C_0 H_{11} O_5 \ (\text{D-glucose residue}) \\ V \ R = H; \ R_1 = C_0 H_5 CO. \end{array}$$

The combination of facts given shows that karatavigenin B is alliogenin benzoate. Since in the NMR spectrum of (IV) in the 5.95-ppm region there is a broad ( $W_{1/2} \approx 20 \text{ Hz}$ ) signal of an axial proton geminal to a benzoate group, and the alkaline hydrolysis of this glycoside forms alliogenin pyranoside (II), it follows that the benzoic acid must esterify the hydroxyl at  $C_2$ . Thus, karatavigenin B is alliogenin 2-O-benzoate, and the glycoside (IV) is its 3-O- $\beta$ -D-glucopyranoside. Alliogenin 3-O-benzoate (V) has been isolated pre-

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<sup>\*</sup> Digit is omitted in Russian original - Publisher.

viously from the same plant [1], and now, in order to distinguish the two isomeric benzoates, it is desirable to call this karatavigenin A.

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